

# MCCRS Alignment for High School Science



Conceptual Academy for  
Foundations of Biology, Biology, Chemistry, Earth and Space  
Science, Physics, Physical Science

## Curriculum Scope and Sequence for MCCRS for Science

This document contains a description of the *Conceptual Academy* High School Program Scope and Sequence for six courses with the corresponding program title in the table below.

Approved Courses for the Secondary Schools of Mississippi	Conceptual Academy Program Name
Foundations of Biology	<i>Conceptual Academy Biology</i>
Biology	<i>Conceptual Academy Biology</i>
Chemistry	<i>Conceptual Academy Chemistry</i>
Earth and Space Science	<i>Conceptual Academy Earth and Space Science</i>
Physical Science	<i>Conceptual Academy Physics and Chemistry Integrated</i>
Physics	<i>Conceptual Academy Physics</i>

The scope and sequence summarizes how the high school program textbook chapters and activities are organized in relation to the MCCRS for Science. The chapter sequencing is designed to build on prior ideas and integrate the three dimensions of the *Framework for K-12 Science Education: Practices, Crosscutting Concepts, and Core Ideas*.

### Course Sequence

The Conceptual Academy High School scope and sequence can ideally progress through the Mississippi Best Practices for CCR Sequencing in Science.

Grade	9	10	11	12
Course	Biology (260131)	Chemistry 1 (400519)	Physics (400820) <u>or</u> Earth and Space (260629)	Physics (400820) <u>or</u> Earth and Space (260629)

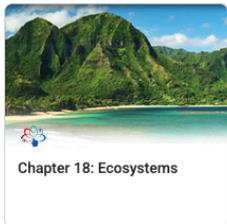
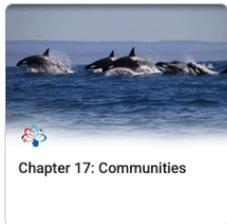
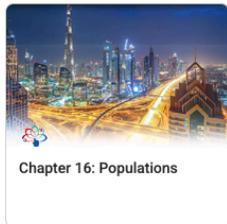
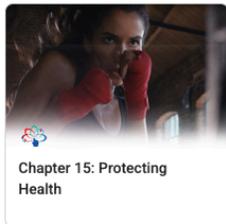
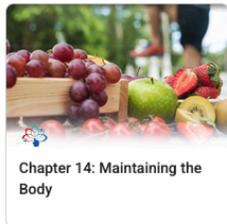
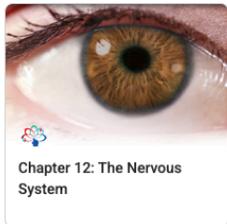
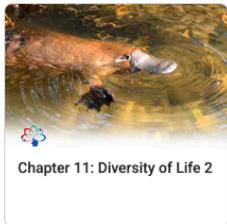
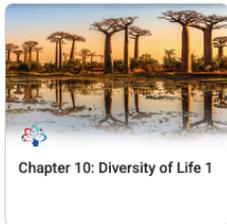
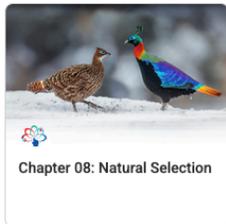
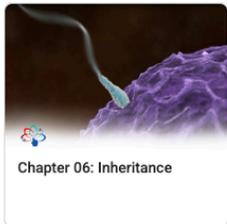
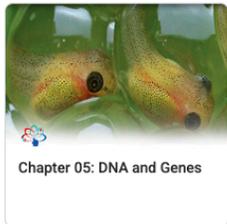
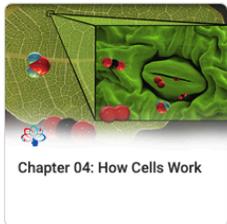
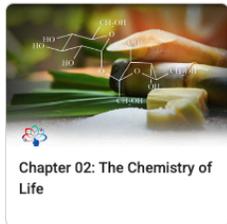
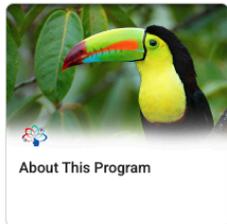
Additionally, the Foundations of Biology program is available for students to gain the basic knowledge needed prior to attempting the rigorous Biology course required for graduation. Combined with the Biology program at 10th grade and the Physical Science program at 11th grade, this sequence would give students the breadth of knowledge across the three core science disciplines in a three-year sequence. If students opted for a fourth year of science they could go into more depth with the Chemistry of Physics programs or gain further breadth with Earth and Space Science.

Grade	9	10	11	12
Course	Foundations of Biology (260628)	Biology (260131)	Physical Science (400700)	Chemistry 1 (400519) <u>or</u> Physics (400820) <u>or</u> Earth and Space (260629)

The order of the standards within each program reflects a purposeful consideration of how to build disciplinary core ideas (DCIs), science and engineering practices (SEPs), and crosscutting concepts (CCCs) through three-dimensional learning, while also maintaining a logical progression through the core content knowledge and covering 100% of the MCCRS for Science.

# Scope and Sequence: Foundations of Biology and Biology

*Conceptual Academy Biology* applies to courses for Foundations of Biology and Biology, with different MCCRS Alignment for each course. Beginning with the chemistry essential to life at the molecular level, we build toward cells, genetics, and inheritance—laying the groundwork for evolution and the remarkable diversity of life. From there, the journey moves into human anatomy and physiology, culminating in an exploration of populations, communities, and ecosystems of which we are an integral part. Throughout, biological concepts connect to real-world examples from medicine, nutrition, health, and biotechnology.

















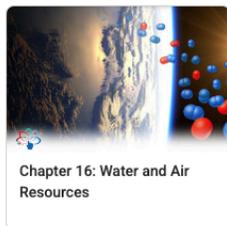
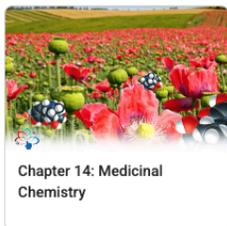
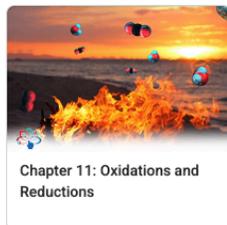
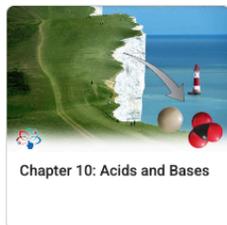
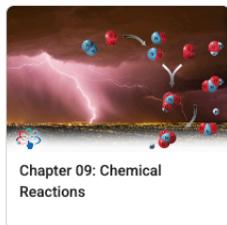
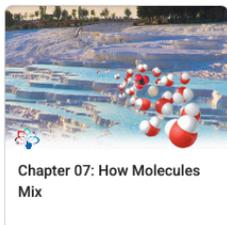
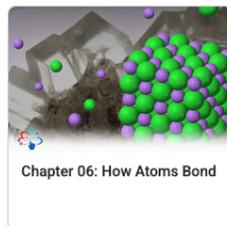
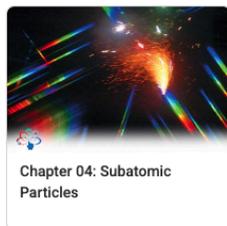
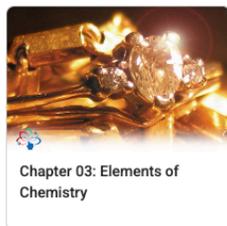
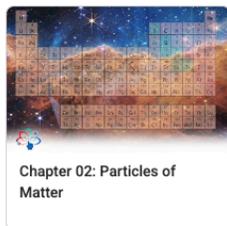
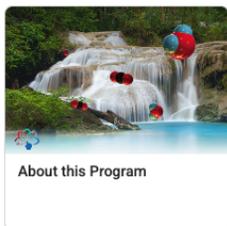






# Scope and Sequence: Chemistry

*Conceptual Academy Chemistry* emphasizes the interconnected ideas that make the molecular world understandable. Beginning with the submicroscopic world of atoms, we explore atomic structure and the periodic table and lay the groundwork for understanding how atoms bond to form molecules. From there, the journey moves into solutions, chemical reactions, and the behavior of acids, bases, and electrons, culminating in the diverse world of organic compounds and polymers. Throughout, chemical concepts connect to real-world examples from materials science, medicine, environmental protection, and energy.





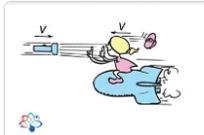
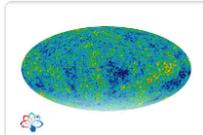






# Scope and Sequence: Earth and Space Science

*Conceptual Academy Earth and Space Science* begins with Earth's deep history and interior structure, we explore plate tectonics, minerals, rocks, and the surface processes that shape our world. From there, the journey moves through oceans, atmosphere, weather, and climate before venturing outward to the solar system, stellar life cycles, galaxies, and cosmology. Foundational physics—from Newton's laws to relativity—is woven throughout.

 About this Program	 ESS 1: The Scientific Perspective	 ESS 2: Earth's History	 ESS 3: Earth's Structure	 ESS 4: Plate Tectonics	 ESS 5: Minerals	 ESS 6: Rocks
 ESS 7: Landforms	 ESS 8: Earth's Waters	 ESS 9: Surface Processes (Part 1)	 ESS 10: Surface Processes (Part 2)	 ESS 11: The Atmosphere	 ESS 12: Understanding Weather and Climate: Moving Air and Water	 ESS 13: Environmental Geology
 ESS 14: Climate Dynamics	 ESS 15: The Solar System	 ESS 16: The Inner Solar System	 ESS 17: The Outer Solar System	 ESS 18: Newton's Laws of Motion	 ESS 19: Gravity	 ESS 20: Atoms, Heat, and Light
 ESS 21: Nuclear Energy	 ESS 22: Stars	 ESS 23: Superstars	 ESS 24: Galaxies	 ESS 25: Relativity	 ESS 26: Cosmology	











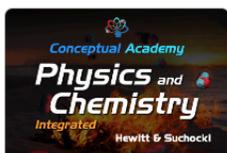






# Scope and Sequence: Physical Science

*Conceptual Academy Physics and Chemistry Integrated* is designed as an introductory-level program for a comprehensive course in physical science. Beginning with Newton's laws of motion, we explore momentum, energy, and heat then progress to electricity, magnetism, and waves. From there, the journey moves into the submicroscopic world of atoms, the periodic table, and how atoms bond to form molecules. We then examine solutions and chemical reactions, culminating in environmental science applications such as water quality, air pollution, climate, and sustainable energy.



About This Program



Chapter 01: About Science



Chapter 02: Newton's First Law



Chapter 03: Newton's Second Law



Chapter 04: Newton's Third Law



Chapter 05: Momentum



Chapter 06: Energy



Chapter 07: Heat



Chapter 08: Electricity



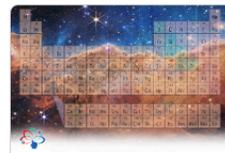
Chapter 09: Magnetism



Chapter 10: Waves and Sound



Chapter 11: Light and Color



Chapter 12: Particles of Matter



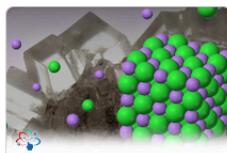
Chapter 13: Elements of Chemistry



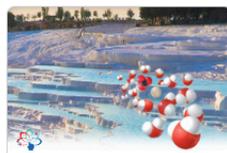
Chapter 14: Subatomic Particles



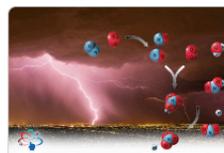
Chapter 15: The Atomic Nucleus



Chapter 16: How Atoms Bond



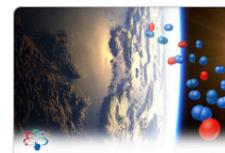
Chapter 17: How Molecules Mix



Chapter 18: Chemical Reactions



Chapter 19: Acid/Base and Redox



Chapter 20: Environmental Science







MCCRS Correlation - Physical Science

Alignment Rating Scale 0 - 2: Minimal (not shown) 2: Partial 3: Moderate 4: Extensive	PHS.1: Nature of Matter						Atomic	PHS.3 Periodic Table					PHS.4: Law of Conserv of Matter and Energy						PHS.5: Newton's Laws of Motion						PHS.6 Waves						PHS.7 Energy				PHS.8 Thermal Energy				PHS.9 Electricity							
	PHS.1.1	PHS.1.2	PHS.1.3	PHS.1.4	PHS.1.5	PHS.1.6	PHS.2.1	PHS.3.1	PHS.3.2	PHS.3.3	PHS.3.4	PHS.3.5	PHS.4.1	PHS.4.2	PHS.4.3	PHS.4.4	PHS.4.5	PHS.4.6	PHS.5.1	PHS.5.2	PHS.5.3	PHS.5.4	PHS.5.5	PHS.5.6	PHS.5.7	PHS.5.8	PHS.6.1	PHS.6.2	PHS.6.3	PHS.6.4	PHS.6.5	PHS.6.6	PHS.6.7	PHS.6.8	PHS.7.1	PHS.7.2	PHS.7.3	PHS.7.4	PHS.8.1	PHS.8.2	PHS.8.3	PHS.8.4	PHS.9.1	PHS.9.2	PHS.9.3	PHS.9.4
13.3 The Periodic Table		3.3						4.0	2.5																																					
13.4 Elements to Compounds		4.0						2.3	2.3																																					
13.5 Naming Compounds			2.3	2.5				3.5	4.0	4.0	4.0																																			
13.6 Most Materials Are Mixtures	4.0	3.5			2.5								2.3																																	
13.7 Classifying Matter	2.3	2.8	4.0	2.7																																										
13.8 The Advent of Nanotechnology																				2.5																										
Chapter 14: Subatomic Particles																																														
14.1 Physical and Conceptual Models	3.5	3.0																																												
14.2 The Electron		2.3		2.3					3.0																																					
14.3 The Atomic Nucleus	2.5	3.0					4.0																																							
14.4 Protons and Neutrons	2.5						4.0	3.0			3.0																																			
14.5 Light Is a Form of Energy													2.3															2.5	3.0			2.7		2.5												
14.6 Atomic Spectra and the Quantum		3.5	2.7				2.3																			3.0		3.5			3.0		2.5	2.7	2.5		2.3									
14.7 Electrons Exhibit Wave Properties							3.0																			2.5	2.5	4.0			2.5		4.0													
14.8 The Shell Model	3.0	2.7					4.0	3.5																																						
14.9 Understanding Periodic Trends		3.7					3.0	3.4	2.7																																					
Chapter 15: The Atomic Nucleus																																														
15.1 Unstable Nuclei																																														
15.2 Radioactivity Is Natural																				2.3																										
15.3 An Imbalance of Forces	3.0						2.5																																							
15.4 Transmutation								3.0					3.0																																	
15.5 Radioactive Half-Life																																														
15.6 Isotopic Dating																																														
15.7 Nuclear Fission		2.3																																												
15.8 Mass and Energy																																														
15.9 Nuclear Fusion		2.3												3.0																																
Chapter 16: How Atoms Bond																																														
16.1 Electron-Dot Structures									2.3																																					
16.2 Ion Formation	3.0	2.3						3.2	3.0																																					
16.3 Ionic Bonds	2.5	3.7						2.3	4.0	2.7	2.5																																			
16.4 Metallic Bonds	3.3	4.0																																												
16.5 Covalent Bonds	2.3	2.7						2.3	4.0		2.5																																			
16.6 Molecular Shape		2.5					2.5		2.3																																					
16.7 Polar Covalent Bonds	3.0	2.5						2.2	3.3																																					
16.8 Molecular Polarity	2.5	2.7							3.0																																					
Chapter 17: How Molecules Mix																																														

MCCRS Correlation - Physical Science

Alignment Rating Scale 0 - 2: Minimal (not shown) 2: Partial 3: Moderate 4: Extensive	PHS.1: Nature of Matter						Atomic	PHS.3 Periodic Table					PHS.4: Law of Conserv of Matter and Energy						PHS.5: Newton's Laws of Motion						PHS.6 Waves						PHS.7 Energy				PHS.8 Thermal Energy				PHS.9 Electricity						
	PHS.1.1	PHS.1.2	PHS.1.3	PHS.1.4	PHS.1.5	PHS.1.6	PHS.2.1	PHS.3.1	PHS.3.2	PHS.3.3	PHS.3.4	PHS.3.5	PHS.4.1	PHS.4.2	PHS.4.3	PHS.4.4	PHS.4.5	PHS.4.6	PHS.5.1	PHS.5.2	PHS.5.3	PHS.5.4	PHS.5.5	PHS.5.6	PHS.5.7	PHS.5.8	PHS.6.1	PHS.6.2	PHS.6.3	PHS.6.4	PHS.6.5	PHS.6.6	PHS.6.7	PHS.6.8	PHS.7.1	PHS.7.2	PHS.7.3	PHS.7.4	PHS.8.1	PHS.8.2	PHS.8.3	PHS.8.4	PHS.9.1	PHS.9.2	PHS.9.3
17.1 Dipole Attractions	3.5	2.3							3.3																																				
17.2 Solutions	4.0	3.3	2.7										2.5																																2.5
17.3 Concentration and the Mole	2.5	2.7		2.5																																									
17.4 Solubility	4.0	2.5	2.7	2.5					2.3		2.3																																		
17.5 How Soap Works	2.5	2.5							2.7																																				
17.6 Softening Hard Water	3.5	2.5						2.3	2.5																																				
17.7 Purifying Drinking Water		2.5											2.3																																
<b>Chapter 18: How Chemicals React</b>																																													
18.1 Chemical Equations														4.0	4.0	2.3																													
18.2 Measuring Molecules	3.0	2.3		2.3					2.3		3.0					3.0																													
18.3 Grams to Moles			2.7						2.7		4.0		3.5	2.5	4.0																														
18.4 Exothermic or Endothermic		2.3						2.3																																4.0					
18.5 Entropy and Chemical Reactions																																													
18.6 Chemical Catalysts		2.3																																											
<b>Chapter 19: Acid/Base and Redox</b>																																													
19.1 Exchanging Protons		2.3								4.0		2.3																																	
19.2 Acid and Base Strength	3.5	2.8	3.0							2.7																															2.5				
19.3 Acidic, Basic, or Neutral		2.3																																											
19.4 Rainwater Is Acidic		3.0								2.7																																			
19.5 Ocean Acidification		2.3								2.3		2.3																																	
19.6 Exchanging Electrons		3.3						2.7				2.5																																	
19.7 Electrochemistry		2.4																																							2.3				
19.8 Types of Batteries		2.8						2.3					3.0																											3.0	2.3	4.0			
19.9 Fuel Cells		2.5																																						2.5	2.5				
19.10 Photovoltaics	3.0	2.3				2.5		3.0																																3.0		4.0			
19.11 Electrolysis		2.5					2.3		2.5	2.5																														2.5		4.0			
19.12 Corrosion and Combustion		3.0						3.0				2.3																																	
<b>Chapter 20: Environmental Science</b>																																													
20.1 Water on the Move																																													
20.2 Water We Consume				2.5																																									
20.3 Polluting Water		2.3																																											
20.4 Earth's Atmosphere	3.0	3.3																																											
20.5 How We Pollute Air																																													
20.6 Global Warming																																													
20.7 Energy through Electricity																																												2.5	2.3
20.8 Fossil Fuels	3.3	2.3																																											
20.9 The Nuclear Industry																																													



# Scope and Sequence: Physics

This program takes a conceptual approach to physics at an introductory level, building on the legacy of Paul Hewitt's acclaimed Conceptual Physics curriculum now tailored specifically for high school students. Beginning with Newton's laws of motion, we explore momentum, energy, and gravity—laying the groundwork for understanding fluid mechanics and heat. From there, the journey moves into electricity and magnetism, waves and sound, and the fascinating behavior of light, culminating in atomic structure and nuclear energy. Physics concepts connect to real-world phenomena and everyday experiences.

